



# Report

Certification of reference material

ERM®-FA005

Poly(styrene) (PS)

Batch No.: **ps980422** 

This material was produced by BASF AG, Ludwigshafen. It can be used for calibrating specific methods investigating polymers. Certified values are the mass averaged molecular weight M<sub>w</sub> by means of light scattering (LS) and the intrinsic viscosity by means of viscometry. Additional, non-certified values are the averaged molecular weights (M<sub>w</sub>, M<sub>n</sub>, M<sub>z</sub>, M<sub>p</sub>) and M<sub>w</sub>/M<sub>n</sub> by means of size exclusion chromatography (SEC). These values are based on results obtained by round robin tests which were initiated and evaluated by the department BAM VI.3. Additional tests that result in non-certified values (IR, NMR, DSC, MFR and determination of density) were exclusively performed in the BAM.

Homogeneity and stability of the material were tested in the BAM, too.

The material has a durability of 15 years for temperatures of +3 °C to +7 °C at maximum.

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#### 1. Abbreviations, symbols and formulas

IR - infrared spectroscopy

NMR - nuclear magnetic resonance spectroscopy

DSC - differential scanning calorimetriy

MFR - melt flow index

SEC - size exclusion chromatography

LS - light scattering

MALLS - multi-angle laser light scattering
LALLS - low-angle laser light scattering

MALDI-TOF-MS - matrix assisted laser desorption/ionisation -

time of flight - mass spectrometry

M<sub>z</sub> - z-averaged molecular weight

D - polydispersity (=  $M_w/M_n$ )

THF - tetrahydrofurane

*x* - mean value

[η] - intrinsic viscosity

$$M_{n} = \frac{\sum_{i=1}^{k} n_{i} * M_{i}}{\sum_{i=1}^{n} n_{i}}$$
 (1)

$$M_{w} = \frac{\sum_{i=1}^{k} n_{i} * M_{i}^{2}}{\sum_{i=1}^{n} n * M_{i}} = \frac{\sum_{i=1}^{k} m_{i} * M_{i}}{\sum_{i=1}^{k} m_{i}}$$
(2)

$$M_{z} = \frac{\sum_{i=1}^{k} n_{i} * M_{i}^{3}}{\sum_{i=1}^{n} n * M_{i}^{2}} = \frac{\sum_{i=1}^{k} m_{i} * M_{i}^{2}}{\sum_{i=1}^{k} m_{i} * M_{i}} = \frac{\sum_{i=1}^{k} z_{i} * M_{i}}{\sum_{i=1}^{k} z_{i}}$$
(3)

#### 2. Introduction

Polymer standards are the basis for calibration of relative methods used for the characterization of molecular weights and weight distribution of polymers. An important method is represented by the Size Exclusion Chromatography (SEC). The polymer, that has to be investigated, will be dissolved in an appropriate solvent and will be separated in columns according to the hydrodynamic radii of macromolecules. These columns are filled with specific gels having various pore sizes and pore distributions. The hydrodynamic volume depends both on the molecular weight and on the structure of dissolved polymers. Therefore, for analysing structurally different polymers various standards are necessary.

The molecular weight of these standards can be measured by means of so-called absolute methods, which do not require any calibration. One of the most important absolute methods is given by measuring the light scattering of a polymer solution. The intensity of the scattered light increases with increasing molecular weight. Apart from determining the refractive indices at different polymer concentrations (refractive index increment), no further information is necessary. For investigations carried out at the BAM a Dawn EOS light scattering photometer (Wyatt) was applied.

The SEC BAM round robin tests were performed and evaluated according to DIN 55 672 – 1. The conditions were obligatory for all participating laboratories. Samples were measured at the BAM using a PL-210 SEC (Polymer Laboratories, Church Stretton, UK). For the calibration of SEC commercially available standards were used (Polymer Standards Service [PSS] GmbH, Mainz) by all participating laboratories. The calculation of the molecular weights was performed using the WINGPC program of PSS, which is based on known mathematical formulas (1) to (3).

As a third method viscometry was used. The determination of the viscosities of polymer solutions with different concentrations and their subsequent extrapolation versus a concentration c=0 results in the so-called intrinsic viscosity [ $\eta$ ]. Applying the equation [ $\eta$ ] = K M<sub>v</sub><sup>a</sup> (K and a are constants available for different solvents and temperatures) a vicosity averaged molecular weight M<sub>v</sub> can be obtained.

These investigations were performed in the BAM according to DIN 51562 – 1 using a AVS/G – Ubbelohde viscometer (Schott, Mainz).

#### 3. List of participating laboratories

Aventis, Frankfurt / M.

Bundesanstalt für Materialforschung und -prüfung, Berlin

Bayer AG, Uerdingen

Bayer AG, Leverkusen

Bundeskriminalamt, Wiesbaden

BMW, Dingolfing

Fraunhofer Institut für Angewandte Polymerforschung, Teltow

RWTH Aachen, Institut für Kunststoffverarbeitung

Institut für Lacke und Farben, Magdeburg

Institut für Polymerforschung, Dresden

Martin-Luther-Universität, Halle-Wittenberg

Max-Planck-Institut für Polymerforschung, Mainz

Polymer Standards Service GmbH, Mainz

Röhm GmbH, Darmstadt

RWTH Aachen, Institut für Textilchemie und Makromolekularen Chemie

Goldschmidt AG, Essen

Technische Universität Dresden

Universität Bayreuth

Universität Erlangen-Nürnberg

Universität Essen

Universität Freiburg

Universität Hamburg, Institut für Technische und Makromolekulare Chemie

Universität Hamburg, Institut für Technische und Makromolekulare Chemie

Universität Leipzig

Johannes-Gutenberg-Universität Mainz, Institut für Makromolekulare Chemie

Johannes-Gutenberg-Universität Mainz, Institut für Physikalische Chemie

Universität Osnabrück

Universität Stuttgart, Institut für Technische Chemie

Universität Stuttgart, Institut für Textil- und Faserchemie

Universität Ulm

Universität - Gesamthochschule Siegen

Viscotek GmbH, Weingarten

#### 4. Synthesis and packing size

The polymer was synthesized by BASF AG, Ludwigshafen. It was filled in 18 brown glass bottles with a volume of 1 litre. Each of these bottles contained ca. 690 g of the polymer. The polymers itself consist of glassy pellets.

The samples were manually splitted. Every participant got approximately 2 grams of the polymeric material.

After certification a certain part of the whole polymer material will be bottled in sizes of 1, 2, 5 or 10 g by the distributor stating the corresponding batch number. The packing procedure will be controlled by the BAM. The remaining part of the material is stored in sealed bottles and can be packed by the distributor if required. The BAM reserves the right to check the packing procedure by taking samples immediately after packing and for an indefinite time.

#### 5. Investigation of homogeneity

In order to separate the uncertainty of the method from the heterogeneity of the sample a multiple measuring of the sample according to ASTM E 826 – 85 is necessary. Since polymer materials are synthezised in batch processes and are repeatedly cleaned by various methods (e.g. re-precipitation) no significant differences were expected a priori.

Overall 16 samples of the polymer (one sample per bottle) were investigated by means of SEC. Every sample was measured twice.

$$M_W = 315400 \pm 600 \text{ g/mol } (\pm 0.20 \%)$$

(confidence interval for 36 values and 95% probability)

Additionally, the statistical accuracy of the SEC method was determined using a polystyrene standard material with a broad polymer distribution. One pellet of the polymer was dissolved in THF. This solution was measured 10 times.

$$M_w = 313300 \pm 400 \text{ g/mol } (\pm 0.14 \%)$$

(confidence interval for 10 values and 95% probability)

The confidence interval of the SEC method is lower than the confidence interval of the homogeneity test. (For comparison: The statistical accuracy of the SEC method according to DIN  $55\,672-1$  has to be at least 2% for  $M_w$ .)

#### 6. <u>Investigation of stability</u>

Stability tests were performed at elevated temperature (40 °C) by storing the polymers for two years. Samples were taken every six month. The molecular weight was determined twice by means of SEC.

Storage time (month)	Molecular weight M <sub>w</sub> (g/mol)
0	319280
6	313750
12	314870
18	315300
24	314320

Only slight changes of 1,73 % (lower than the uncertainty of the SEC method) were detected within the timeframe. The results are within the statistical accuracy according to DIN 55 672 - 1.

## 7. Non-certified values

**NMR-spectroscopy:** intensity ratio of aliphatic and aromatic

protons: 3 / 5, no determination of end-groups

**IR-Spectroscopy:** IR-spectrum corresponds with reference

spectra

**Differential Scanning Calorimetry:** glass transition temperature  $T_g = 103.8$  °C

Melt Flow Index: 1.48  $\pm$  0.02 g/10 min (5 kg, 200 °C, according

to DIN ISO 1133)

**Density:** 1.03 g/ml (25 °C, according to DIN 53479)

## 8. Results of the round robin tests

## **Non-certified values**

# 1. Averaged mol. weights ( $M_w$ , $M_n$ , $M_z$ and $M_p$ ) and polydispersity $M_w/M_n$ by size exclusion chromatography (SEC)

	Mean values of investigators				
Investigator	Weight-	Weight- Number- Z-average Mol. weight			
	average	average		at peak max.	$M_w/M_n$
	M <sub>w</sub> [g/mol]	M <sub>n</sub> [g/mol]	M <sub>z</sub> [g/mol]	M <sub>p</sub> [g/mol]	
1	30900ß	106900	537600	248200	2.89
2	316000	139000	537000	294000	2.27
3	299800	113300	527700	239300	2.65
4	320750	142900	577150	239650	2.24
5	298750	138950	519150	242450	2.15
6	297800	133200	508900	250900	2.24
7	301300	149300	487600	-	2.02
8	313550	147500	539600	279000	2.13
9	320900	139300	560000	267850	2.30
10	322450	160700	539000	306800	2.01
11	316500	158000	551000	253550	2.00
12	328150	143100	561650	319800	2.29
13	304550	138700	544800	267000	2.20
14	324150	139800	552950	-	2.32
15	303000	143000	532000	257000	2.12
Mean values	311800	139600	539000	266700	2.25
Confidence interval	4700	6400	10000	11800	0.11
[ %]	1.42	4.54	1.86	4.44	4.76

## **Certified Values**

# 2. Weight-average molecular weight (Mw) by light scattering (LS)

	Mean values of investigators		
Investigator	Weight-average molecular weight M <sub>w</sub> [g/mol]		
1	359000 b)		
2	342000 b)		
3	394900 a)		
4	338300 <sup>c)</sup>		
5	329000 b)		
6	362000 <sup>c)</sup>		
7	338700 b)		
8	334600 b)		
9	364300 a)		
10	369600 b)		
11	324600 <sup>c)</sup>		
12	345400 <sup>c)</sup>		
13	351900 <sup>d)</sup>		
Mean value	349800		
Confidence			
interval	9700		
[ %]	2.77		

# 3. Intrinsic viscosity by viscometry

Investigator	Mean values of investigators Intrinsic viscosity		
Investigator			
	[η] [ml/g]		
1	103.43 a,b)		
2	100.19 a,b)		
3	107.21 <sup>a,b)</sup>		
4	105.17 <sup>a,b)</sup>		
5	107.86 a,b)		
6	102.84 <sup>c)</sup>		
7	103.20 <sup>c)</sup>		
8	104.35 <sup>c)</sup>		
Mean value	104.28		
Confidence interval	2.30		
[ %]	2,20		

#### **Experimental conditions**

- 1) The experimental conditions were determined by the DIN 55 672 1 (GPC using tetrahydrofurane (THF) as eluent.
- <sup>2)</sup> Values correspond to a Rayleigh-ratio  $R_{\Theta} = 1,406$  E-5 cm<sup>-1</sup> at 633 nm in toluene
  - a) Low-Angle Laser Light Scattering (LALLS), b) Multi-Angle Laser Light Scattering (MALLS), c) Size Exclusion Chromatography coupled with MALLS-Detector,
  - <sup>d)</sup> Size Exclusion Chromatography coupled with Right-Angle Laser Light Scattering (RALLS) Detector

Investigator	Method	Angle (°)	Solvent	Equipment	Wave length (nm)	dn/dc
1	MALLS	30-150	THF	Dawn EOS	690	0.1840
2	MALLS	30-150	Toluol	Dawn EOS	690	0.1096
3	LALLS	6-7	THF	KMX-6	633	0.1840
4	SEC-LS	30-145	THF	Dawn F	488	0.1990
5	SEC-LS	30-145	THF	Dawn F	633	0.1840
6	MALLS	20-145	THF	Fica (SLS)	633	0.1840
7	MALLS	30-150	THF	Dawn DSP	488	0.1990
8	MALLS	30-145	Toluol	FIKA 50	633	0.1096
9	LALLS	6-7	THF	KMX-6	633	0.1840
10	MALLS	30-145	THF	Sofica	633	0.1840
11	SEC-LS	30-150	THF	Dawn F	633	0.1840
12	SEC-LS	30-145	THF	Dawn F	633	0.1840
13	SEC-LS	90	THF	TDA-300	633	0,1840

<sup>&</sup>lt;sup>3)</sup> In THF at 30 °C, 6 concentrations from 1 to 5 g/l in an Ubbelohde type viscometer according to HUGGINS <sup>a)</sup> and KRÄMER <sup>b)</sup> following DIN 51562-1, resp. by means of a capillary viscometer (Viscotek, Weingarten) <sup>c)</sup>

### 9. References

- DIN 55 672 1 (GPC using tetrahydrofurane (THF) as eluent)
- DIN 55 672 2 (GPC using N,N Dimethylacetamide (DMAC) as eluent)
- DIN 51 562 1 (Viscometry: Determination of kinematic viscosity using a
   Ubbelohde Viscometer, Part1: Design and realisation of measurements
- BAM VI. 301 standard working procedure (StAA 7.2.5.1.) (GPC using THF as eluent)
- BAM VI. 301 StAA 7.2.5.2. (determination of the molecular weight of polymers using LALLS (Low-Angle Laser Light Scattering))
- BAM VI. 301 StAA 7.2.5.3. (determination of the viscosity of polymers)